

Ha

# Complete CHEMISTRY

## IIT-JEE · NEET · CBSE eBOOKS CLASS 11&12th



## CLASS 11th IUPAC Nomenclature

## **01.** Format for IUPAC Name

$2^{\circ}$ – prefix	$1^{\circ}$ – prefix	word root +	$1^{\circ}$ – suffix	+ $2^{\circ}$ - suffix
Substituents with locants	Cyclo	Alk word according to carbon in parent C chain	<ul> <li>ane</li> <li>ene</li> <li>yne</li> <li>diene</li> <li>triene</li> <li>diyne</li> </ul>	According to main functional group given in priority table

#### (i) Locant :

- Locants are seprated by (,) comma.
- Locants and alphabates are separated by hyphen (-). [2, 3 dimethyl pentane]
- cis, trans, sec., tert and meso are separated by hyphen (-) [cis-2-butene]
- di, tri, iso, neo and cyclo are neither separated by comma nor by hyphen

### (ii) Prefix :- According to substituents

2° prefix

2-methyl

Prefix (es) are written in alphabetical order before root word.

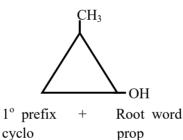
Prefix  $\leftarrow \begin{bmatrix} 1^{\circ} & \text{or } p - prefix \\ 2^{\circ} & \text{or sec.} - prefix \end{bmatrix}$ 

Cyclo is 1° prefix and used for cyclic compound.

2° prefix is used for substituents and written before 1° prefix.

For cyclic compounds :  $2^{\circ}$  prefix +  $1^{\circ}$  prefix + Root word +  $1^{\circ}$  suffix +  $2^{\circ}$  suffix

### **Example**



1° suffix 2° suffix ol

ane

Substituents	Prefix	Substituents	Prefix
— R	Alkyl group	— OR	Alkoxy
— X (F, Cl, Br, I)	Halo	-N <sup>\$0</sup>	Nitro
-0 - N = 0	Nitrite	-N = O	Nitroso
— CH <sub>2</sub> OH	Hydroxy methyl	— CH <sub>2</sub> Cl	Chloro methyl
— NHC <sub>2</sub> H <sub>5</sub>	Ethyl amino		

For acyclic compounds :  $2^{\circ}$  prefix + Root word +  $1^{\circ}$  suffix +  $2^{\circ}$  suffix.

e-Learning for IIT-JEE | NEET | CBSE | 9/10

3

## **IUPAC** Nomenclature

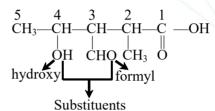
Number of carbons	Root word	Number of carbons	Root word	Number of carbons	Root word
1	Meth	6	Hex	11	Undec
2	Eth	7	Hept	12	dodec
3	Prop	8	Oct	13	tridec
4	But	9	Non	20	Eicos
5	Pent	10	Dec		

(iii) Root word : According to number of carbons in parent C-chain.

(iv) Primary suffix : According to saturation and unsaturation.

$C - C \longrightarrow$ ane	$2C = C \longrightarrow diene$
$C = C \longrightarrow ene$	$2C \equiv C \longrightarrow diyne$
$C \equiv C \longrightarrow yne$	$3C = C \longrightarrow triene$

(v) Suffix : According to senior most of F. G.



3-formyl-4-hydroxy-2-methyl pentanoic acid

## 02. IUPAC Rules

- (a) Selection of longest continuous parent carbon chain
- (b) Numbering in selected parent carbon chain.
- Selection of longest continuous parent C chain :

Sub rule (i) : Selection of longest continuous parent C – chain containing functional group or multiple bond or substituents.

**Priority order :** Functional group > Multiple bond > Substituents

Example

Sub rule (ii) : If carbon containing functional group is present then include carbon of that functional group in parent chain.

Example

$$CH_{3} - \begin{bmatrix} C - CH \\ | \\ CH_{2} \end{bmatrix} \begin{bmatrix} C \\ | \\ CH_{2} \end{bmatrix} = CH_{3}$$

$$CH_{3} - CH - CH_{2} - C - CH_{2} - CH_{2}$$

$$CH_{3} - CH - CH_{2} - CH_{2} - CH_{2}$$

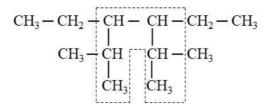
$$CH_{3} - CH - CH_{2} - CH_{2} - CH_{2}$$

e-Learning for IIT-JEE | NEET | CBSE | 9/10th

4

Sub rule (iii) : If more than one equal chains of carbon are possible then select one which containing maximum number of substituents.

Example



Sub rule (iv) : If more than one multiple bonds are present then select one which containing maximum number of multiple bonds.

Example  $\begin{array}{c} \hline CH_2 = CH - CH - CH_2 - CH_2 - CH_3 \\ | \\ CH = CH_2 \end{array}$ 

(a) Numbering in selected parent carbon chain :
 Sub rule (i) : Selected parent carbon chain is numbered from that side from which functional group or multiple bond or substituents gets lowest number.

**Priority order :** Functional group > Multiple bond > Substituents

Example

$$\begin{array}{c} {}^{4}_{CH_{3}} - {}^{3}_{CH} - {}^{2}_{CH_{2}} - {}^{1}_{CH_{3}} \\ {}^{0}_{OH} \\ {}^{5}_{OH} \\ {}^{5}_{CH_{2}} - {}^{4}_{CH_{2}} - {}^{3}_{CH} - {}^{2}_{CH} = {}^{1}_{CH_{3}} \\ {}^{1}_{OH} \\ {}^{1}_{OH} \\ {}^{2}_{OH} \\ {}^{2}_{OH} \\ {}^{1}_{OH} \\ {}^{2}_{OH} \\$$

Sub rule (ii) : If carbon containing functional group is present then give lowest possible number to carbon of that functional group.

Example

5

$$CH_3 - CH_2 - \overset{2}{CH} - \overset{3}{C} - CH_2 - CH_3$$
  
 $\downarrow \qquad \downarrow \\ C \equiv N \quad CH_2 \\ 1 \quad H_2 \\ C \equiv N \quad CH_2 \\ 0 \\ CH_3 - \overset{2}{C} - \overset{3}{C} - \overset{4}{C} H_2 - \overset{4}{C} H_3 \\ 0 \\ CH_3 - \overset{2}{C} - \overset{3}{C} H_2 - \overset{4}{C} H_3 \\ 0 \\ CH_3 - \overset{2}{C} - \overset{3}{C} H_2 - \overset{4}{C} H_3 \\ 0 \\ CH_3 - \overset{2}{C} - \overset{3}{C} H_2 - \overset{4}{C} H_3 \\ 0 \\ CH_3 - \overset{2}{C} - \overset{3}{C} H_2 - \overset{4}{C} H_3 \\ 0 \\ CH_3 - \overset{2}{C} - \overset{3}{C} H_2 - \overset{4}{C} H_3 \\ 0 \\ CH_3 - \overset{2}{C} - \overset{3}{C} H_2 - \overset{4}{C} H_3 \\ 0 \\ CH_3 - \overset{2}{C} - \overset{3}{C} H_2 - \overset{4}{C} H_3 \\ 0 \\ CH_3 - \overset{2}{C} - \overset{3}{C} H_2 - \overset{4}{C} H_3 \\ 0 \\ CH_3 - \overset{2}{C} - \overset{3}{C} H_2 - \overset{4}{C} H_3 \\ 0 \\ CH_3 - \overset{2}{C} - \overset{3}{C} H_2 - \overset{4}{C} H_3 \\ 0 \\ CH_3 - \overset{2}{C} - \overset{3}{C} H_2 - \overset{4}{C} H_3 \\ 0 \\ CH_3 - \overset{2}{C} - \overset{3}{C} H_2 - \overset{4}{C} H_3 \\ 0 \\ CH_3 - \overset{2}{C} - \overset{3}{C} H_2 - \overset{4}{C} H_3 \\ 0 \\ CH_3 - \overset{2}{C} - \overset{3}{C} H_2 - \overset{4}{C} H_3 \\ 0 \\ CH_3 - \overset{2}{C} - \overset{3}{C} H_2 - \overset{4}{C} H_3 \\ 0 \\ CH_3 - \overset{2}{C} - \overset{3}{C} H_2 - \overset{4}{C} H_3 \\ 0 \\ CH_3 - \overset{2}{C} - \overset{4}{C} H_3 \\ 0 \\ CH_3 - \overset{4}{C} H_3 \\ CH_3 \\ CH_3 \\ CH_3 - \overset{4}{C} H_3 \\ CH_3 \\ C$ 

Sub rule (iii) : Only for symmetrical conditions

(a) When only two substituents are present at symmetrical position then follow alphabetical order.

